

Section 2 3 Carbon Compounds Answer Key

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Section 2-3 Carbon Compounds. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. erinwd. carbon compounds flashcards. Terms in this set (76) What is organic chemistry the study of? the study of all compounds that contain bonds between carbon atoms. What is a monomer? small unit that can join together with other ...

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Section 2-3: Carbon Compounds Questions and Study Guide ...

Section 2-3 Carbon Compounds This section explains how the element carbon able to form millions of carbon, or organic, compounds. It describes the four groups of organic compounds found in living things. The Chemistry of Carbon How many valence electrons does each carbon atom have? Each carbon atom has four electrons.

Section 2-3 Carbon Compounds

44 Chapter 2 1 FOCUS Objective 2.3.1 Describe the functions of each group of organic compounds. Vocabulary Preview As students read, have them make a concept map using the section's Vocabulary terms, excluding the words

Section 2-3 2-3 Carbon Compounds

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Biology Section 2.3 Carbon Compounds. Test on Chapter 2 - 10/13/2015. STUDY. PLAY. What is the definition of organic and inorganic chemistry? organic chemistry - the study of compounds that contain bonds between carbon atoms inorganic chemistry - the study of all other compounds.

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EdI

Chapter 2.3: Carbon Compounds (1) TRIGLYCERIDES - provide insulation that keeps you warm while protecting your internal organs with a layer of padding. (2) STEROIDS - include hormones and cholesterol. Cholesterol is produced by the body and consumed through food, and it... (3) PHOSPHOLIPIDS - ...

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section 2-3 Carbon Compounds 1. Ribonucleic acid (RNA) 2. Deoxyribonucleic acid (DNA)

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Section 2-3 Carbon Compounds (pages 44-48) Key Concept • What are the functions of each group of organic compounds? The Chemistry of Carbon (page 44) 1. How many valence electrons does each carbon atom have? 2. What gives carbon the ability to form chains that are almost unlimited in length? Macromolecules (page 45) 3.

Section 2-3 Carbon Compounds - Hazleton Area High School

Any of the hydrogen atoms can be replaced with another carbon atom covalently bonded to the first carbon atom. In this way, long and branching chains of carbon compounds can be made (Figure 2.13 a). The carbon atoms may bond with atoms of other elements, such as nitrogen, oxygen, and phosphorus (Figure 2.13 b).

2.3 Biological Molecules - Concepts of Biology - 1st ...

Section 2.3 Covalent Compounds. Covalent Compounds. ... CS₂. First word in compound name: carbon N₂O₄. First word in compound name: dinitrogen 2. The second word in the compound name is the name of the second element in the formula . that has been changed to end with - ide. In all cases, use a prefix (Table 2.3.2) to indicate

Section 2.3 Covalent Compounds

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Carbon - Carbon - Compounds: More than one million carbon compounds have been described in chemical literature, and chemists synthesize many

new ones each year. Much of the diversity and complexity of organic forms is due to the capacity of carbon atoms for bonding with one another in various chain and ring structures and three-dimensional conformations as well as for linking with other atoms.

Carbon - Compounds | Britannica

solid carbon and water vapor.) What is different? (In one case, heating causes the change; in the other, the addition of an acid.) You can also compare the electrolysis of water to the distillation of water (Section 2.2). 1 L2 L2 2 L1 Section Resources 48 Chapter 2 2.3 Elements and Compounds Guide for Reading Key Concepts • How are elements and

2.3 Elements and Compounds

Section 2-3: Carbon Compounds Four groups of carbon compounds found in living things are carbohydrates, lipids, nucleic acids, and protein. Living things use carbohydrates as their main source of energy. Plants and some animals also use carbohydrates for structural purposes. Lipids can be used to store energy.

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