

## Read Free Definition For Saturated Solution

# Definition For Saturated Solution

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### **Definition For Saturated Solution**

A saturated solution is a chemical

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solution containing the maximum concentration of a solute dissolved in the solvent. The additional solute will not dissolve in a saturated solution. The amount of solute that can be dissolved in a solvent to form a saturated solution depends on a variety of factors.

### **Saturated Solution Definition and**

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## **Examples**

A saturated solution is a solution in which there is so much solute that if there was any more, it would not dissolve. When a saturated solution is placed in contact with additional solute, solute neither dissolves nor is deposited. When water, or any solvent, has dissolved as much of any substance as it

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can, it is a saturated solution.

### **Saturated solution definition and meaning | Collins ...**

According to the literature data, the pH of the saturated solution of Ca [ (OH).sub.2] is about 12.4, used as a standard of highly alkaline pH, which well corresponds to the results of the

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present research. Modelling the solubility of sparingly soluble compounds depending on their particles size

### **Saturated solution - definition of saturated solution by ...**

saturated solution definition: 1. a solution (= a liquid containing a solid) in



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which as much solid as possible is dissolved 2. a....

### **SATURATED SOLUTION | meaning in the Cambridge English ...**

A saturated solution is one where there are about equal amounts of particles or solutes and solvent in the solution. If you live on one of the coasts, you've

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probably gone to the beach and played...

### **Saturated Solution: Definition & Examples - Video & Lesson ...**

A saturated solution is a solution that contains the maximum amount of solute that can be dissolved under the condition at which the solution exists. In

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chemistry, after studying solutions and properties of the solution, one can understand that a solution can reach a status of saturation.

## **What is a Saturated Solution - Preparation, Types & Examples**

The term saturated solution is used in chemistry to define a solution in which

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no more solute can be dissolved in the solvent. It is understood that saturation of the solution has been achieved when any additional substance that is added results in a solid precipitate or is let off as a gas.

### **Examples of Saturated Solution**

Saturated definition is - full of moisture :

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made thoroughly wet. How to use saturated in a sentence.

## **Saturated | Definition of Saturated by Merriam-Webster**

A saturated solution is a solution that contains the maximum amount of solute that is capable of being dissolved. At 20°C, the maximum amount of NaCl that

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will dissolve in 100. g of water is 36.0 g. If any more NaCl is added past that point, it will not dissolve because the solution is saturated.

### **Saturated and Unsaturated Solutions | Chemistry for Non-Majors**

The air was saturated by it just as water

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may hold a chemical in solution. Distinguished from the Patrician, who was a saturated solution. Then he smelled the strange fabric, saturated with the man-smell. Even for this his travelled lordship, seasoned and saturated, had no laugh.

### **Saturated Synonyms, Saturated**

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## **Antonyms | Thesaurus.com**

A supersaturated solution is a solution that contains more than the average solvent that can be dissolved at a given temperature. The recrystallization of the excess dissolved solvent in a supersaturated solution can be started by inserting a tiny solute crystal, called a seed crystal. What is a supersaturated



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sugar solution?

## **Supersaturated Solution - Definition, Examples ...**

saturated solution (Noun) a solution in which the solvent can dissolve no more of a specific solute at a particular temperature How to pronounce saturated solution?

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## **What does saturated solution mean?**

A supersaturated solution is a solution with more dissolved solute than the solvent would normally dissolve in its current conditions. Supersaturation is achieved by dissolving a solute in one set of conditions, then transferring it to

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other conditions without triggering any release of the solute.

### **What Is a Supersaturated Solution?**

solution in which the maximum amount of solvent has been dissolved. Any more solute added will sit as crystals on the bottom of the container. A B C D E F G H I J K L M N O P Q R S T U V W X Y Z

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## **saturated solution definition - Northwestern University**

saturated solution A solution that contains all of a substance capable of dissolving; a solution of a substance in equilibrium with an excess undissolved substance. You will also like... Movement of Molecules Across Cell Membranes

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## **Saturated solution Definition and Examples - Biology ...**

A saturated solution contains more solute per volume of solvent than an unsaturated solution. The solute has dissolved until no more can, leaving undissolved matter in the solution. Usually, the undissolved material is

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denser than the solution and sinks to the bottom of the container.

### **What Is an Unsaturated Solution in Chemistry?**

A solution is a mixture of two or more substances where one substance is completely dissolved in the other. The substance that is dissolved is called the

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solute and the substance that does the ...

### **What is the difference between Solubility and a Saturated ...**

From. Wikipedia. This example is from Wikipedia and may be reused under a CC BY-SA license. It is generally used as a saturated aqueous solution which is

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diluted to the required concentration, expressed as a percentage concentration of the saturated solution (a 100% solution).

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