

## Answers For Chemistry Module 13

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### Answers For Chemistry Module 13

Answers for the objective-response questions and marking guidelines for the short-answer questions are also provided The sample questions, sample answers and marking guidelines CH12-13 2-6 Module 7 Organic Chemistry Question Marks Content Syllabus Outcomes Bands Mod 7 - 1\* 1 Mod 7 Nomenclature CH12-7, CH12-14 2-3

### [EPUB] Answers For Chemistry Module 13

swansonfamily TEACHER. APOLOGIA CHEMISTRY MODULE 13. enthalpy. product. reactant. change in enthalpy. The heat content of a system at constant pressure. when a reaction is exothermic, energy will be thought of as th.... when a reaction is endothermic, energy will be thought of as t....

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In order to answer any question about the energy involved in a chemical reaction, we must first get  $\Delta H$ . Since Table 13.2 contains all needed info., we will use Hess's Law:  $\Delta H^\circ = (2 \text{ moles}) \times (-45.9 \text{ kJ/mole}) + (4 \text{ moles}) \times (-242 \text{ kJ/mole}) - (2 \text{ moles}) \times (33.2 \text{ kJ/mole})$ .  $\Delta H^\circ = -1,126 \text{ kJ}$ .

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Solutions and Tests for Exploring Creation With Chemistry ANSWERS TO THE REVIEW QUESTIONS FOR MODULE #13 I. Potential energy is stored in the bonds of the molecules in the reaction. Kinetic energy is the heat that is either released or absorbed in the reaction. 2.

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energy level is equal to the square of the principle quantum number of that energy level \_\_\_\_12  
The

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